Egg Density Experiment

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# Scientific Method

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### Step 1. Research Question

### Can an egg float in saltwater?

# Step 2. Research

- Density is a measure of mass per volume. Density is a measure of how compact the mass in a substance or object is.
- Mass is how much stuff something is made of (Kg, g)
- Volume is how much space an object takes up (L, ml)
- Water density: 1 g per mL (ocean water: salt makes the water denser)
- Egg density: 1.03 g/mL
- Oil density: 0.93g/cm3
- Wood density: about 1.5 g/cc
- Rock density: 3.7g/cm3

"the density of the egg is higher than the density of fresh water"



### Step 3. Hypothesis

### The egg will float in saltwater. (imagine ocean!)

# Step 4. Experiment

- Supplies Needed: 3 eggs, 3 tall drinking glasses, salt, 1 Tablespoon
- Step1. Fill a tall glass about 2/3 full of water and carefully place th e egg
- Step 2. Fill another glass about 2/3 full of water and add 1 Tablesp oon (T) of salt to the water and stir until it is completely dissolved. T hen place the egg carefully
- Step 3. Add one more T of salt to the water. Then stir until it is com pletely dissolved.
- Step 4. Fill another glass about ½ full of water and add 2 T of salt a nd stir. Then place the egg carefully. Next, add water to the top o f the floating egg

# Step 5. Observation

#### Fresh water



#### Adding 1 T of salt



# Step 5. Observation

#### Adding 2 T of salt



#### Adding fresh water again



### Step 6. Results

When an egg is placed in fresh water, it sinks to the bottom of the glass it is placed in because the egg density is highe r than water density

When you add salt to the water, the egg floats because y ou are increasing the density of the water

Water density < Egg density : the egg sinks</li>
Water density > Egg density : the egg floats
Water density = Egg density : the egg stays in the middle

### Step 7. Share my results

### Take it further:

Instead of egg, you can use oil, pencils, rock, bread, etc., and see if they sink or float in fresh water/saltwater